

## Article

### Hydraulic Activity and Microstructure Analysis of High-Titanium Slag

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**Abstract:** To explain the relationship between the hydration activity of high-titanium slag and its microstructure, the hydration activity of high-titanium slag was determined, then the mineral phase and microstructure characteristics of high-titanium slag glass phase and blast furnace slag were investigated using a series of analytical methods, which contain X-Ray Diffraction (XRD), Scanning Electronic Microscope (SEM), Fourier Transform Infrared spectroscopy (FTIR), Raman spectroscopy, and Nuclear Magnetic Resonance spectroscopy (NMR). The results showed that in slow-cooled high-titanium slag, the hydration inert mineral content was about 98%, and the glass phase content was less than 2%, hence, the hydration activity of slow-cooled high titanium slag accounted for less than 25% of that of the blast furnace slag. The content of the glass phase in water-quenched high-titanium slag was 98%, but the microstructure of the glass phase was very different from that of the blast furnace slag. The glass phase of high-titanium slag has stable forms, which are  $\text{TiO}_4$  monomers, chain or sheet units  $\text{O-Ti-O}$ , and a small amount of 6-coordination  $\text{Ti}^{4+}$ . The Ti makes the  $\text{SiO}_4$  tetrahedron in a glass phase network not only a monosilicate, but more stable forms of disilicates and chain middle groups also appeared. The relative bridge oxygen number increased to 0.2, hence, the hydration activity of water-quenched high-titanium slag took up less than 37% of that of the blast furnace slag.

**Keywords:** high-titanium slag; glass phase; mineral phase; microstructure; Raman spectroscopy

## 1. Introduction

High-titanium slag is a kind of granular or massive waste slag that is produced by the quenching or natural cooling of molten slag discharged from the pig iron obtained via smelting vanadium titanium magnetite [1]. Hydraulicity refers to the property that a material can be hardened in humid air and water to form a stable compound after being ground into a fine powder and mixed with water to form a slurry, also known as hydraulic activity, which is the most basic property of cementitious material. Because the content of  $\text{TiO}_2$  in high-titanium slag is about 20%, and because the slag has low latent hydraulic activity (simply termed as hydraulic activity, also known as pozzolanic activity), it cannot be widely used as a building material. Therefore, it is stored in slag yards, thus occupying a lot of land and causing the waste of titanium resources [2]. At present, 80 million tons of high-titanium slag have accumulated in China, and it is still increasing at a rate of 3 million tons per year, and the utilization rate is less than 3% per year [3–7]. Chemical and mechanical activation have been used to improve the hydraulic activity of high-titanium slag, but these methods have not had a significant effect. Su et al. compared the effects of several alkali activators. When the mix proportion of the high-titanium slag was 70% in the cementitious materials, and the mix proportion of the activator was

4% in the cementitious materials, the best result increased the compressive strength of the cement mortar after 28 d from 6.5 MPa to 8.0 MPa, which is an increase of only 1.5 MPa [8]. Yang et al. studied the characteristics of grinding activated high-titanium slag. When the mix proportion of the high-titanium slag was 30% in the cementitious materials, and its specific surface area was in the range of 300–500 m<sup>2</sup>/kg, the compressive strength of cement after 28 d did not increase monotonically with the fineness, but there is a suitable fineness value of 400 m<sup>2</sup>/kg for the highest compressive strength. The compressive strength of the cement after 28 d was only 23 MPa, which still indicates a very low hydraulic activity [9].

The hydration characteristics of slag are closely related to its hydraulicity. Ao compared the hydration products of blast furnace slag and high-titanium slag, and found that there was no significant difference in the types and morphology of the hydration products of the two kinds of slag cement, except that the amount of high-titanium slag hydration products was small and the C–S–H gel crystallinity was poor and had an irregular rolled shape, and the strength of the cement specimen was lower due to lower density [10]. Shi et al. studied the hydration behavior and hydration products of crystalline minerals in slow-cooled high-titanium slag. It was thought that the existence of elemental Ti promoted the formation of hydration inert minerals such as perovskite and titanopyroxene, which is the main reason for the low hydraulic activity of the slow-cooled high-titanium slag [11]. In fact, even after water quenching treatment, the hydraulic activity of high-titanium slag was still lower and the setting time longer than that of blast furnace slag. The content and microstructure of the glass phase were the key factors affecting the hydraulic activity of slag [12–14]. What is the content of glass phase in high-titanium slag? Does the low content of glass phase restrict the hydraulic activity of high-titanium slag? What are the differences between the glass phase microstructure of high-titanium slag and that of blast furnace slag, and how do these differences affect the hydraulic activity of the glass phase?

In this study, the hydraulic activities of high-titanium slag and blast furnace slag are compared. Then, the mineral phase characteristics and the microstructure characteristics of the glass phase of high-titanium slag and blast furnace slag are investigated contrastively, using a series of analytical methods such as XRD, SEM, FTIR spectroscopy, Raman spectroscopy, and NMR spectroscopy. The inevitable connection between the glass phase microstructure and the hydraulic activity is also revealed, with a view to providing new theoretical guidance for improving the hydraulic activity of high-titanium slag.

## 2. Experiment

### 2.1. Raw Materials

Three kinds of slag were used in the experiment. The slow-cooled high-titanium slag was procured from Panzhihua Iron and Steel Corporation (Panzhihua, China) and was labeled TM. The water-quenched high-titanium slag was procured from Tranvic Group (Chengdu, China) and was labeled TS. In addition, the blast furnace slag was procured from Beijing Shougang Group (Beijing, China), and was labeled PS. The chemical composition (mass fraction) of the three types of slag was determined according to GB/T 176—2017, the measurement results were within the error range specified in this standard, listed in Table 1.

**Table 1.** Chemical composition of slag (wt%).

Sample	CaO	SiO <sub>2</sub>	Al <sub>2</sub> O <sub>3</sub>	TiO <sub>2</sub>	MgO	Fe <sub>2</sub> O <sub>3</sub>	Na <sub>2</sub> O	LOI	Sum
TM	26.64	24.76	13.22	20.39	8.37	1.20	0.95	1.64	97.17
TS	28.36	26.36	13.33	17.18	8.39	1.31	0.16	1.30	96.39
PS	35.92	33.65	16.90	0.07	10.13	0.13	0.33	1.55	98.67

## 2.2. Instruments

An X-ray diffractometer (D/Max 2200, Japan, Cu K ray with  $\lambda = 0.15418$  nm) was used for X-ray diffraction. A field emission scanning electron microscope (Hitachi S4800, Tokyo, Japan) equipped with an energy dispersive X-ray spectrometer (EDS) was used in this study. In addition, a micro-infrared spectrometer (Bruker VERTEX70 from Cernit Co., Ltd. Germany), having a frequency band range of 400 to 4000  $\text{cm}^{-1}$  and a resolution of 2  $\text{cm}^{-1}$ , was used for infrared spectroscopy. A laser Raman spectrometer (Horiba HR Evolution, Japan) was used for Raman spectroscopy. The experiment was conducted at room temperature with a laser wavelength of 532 nm and within a frequency band of 100–2000  $\text{cm}^{-1}$ . Moreover, a fully digital NMR spectrometer (AVANCE 400 (SB) from Bruker Bio-spin, Switzerland) was used for NMR spectroscopy. The resonance frequencies of  $^{29}\text{Si}$  and  $^{27}\text{Al}$  were 79.49 Hz and 104.23 Hz, respectively.

## 2.3. Test Method

### 2.3.1. Hydraulic Activity Test

(1) According to Appendix A of GB/T 18046—2017, the compressive strength ratio of the prepared 50% slag powder sample and Portland cement reference samples was taken as the index of hydraulic activity  $A$  of their age. (2) The content of active  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  dissolved in saturated limewater was determined by the method of boiling circumfluence [15]. The percentage of the amount of  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  dissolved to the total amount of  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  is the pozzolanic activity rate  $K_a$  of the slag.

### 2.3.2. Content of Glass Phase

(1) According to Appendix C of GB/T 18046—2017, Jade software was used to fit the XRD patterns of the slag. The ratio of the crystal diffraction peak area to the total diffraction peak area is slag crystallinity, and 100% minus the crystallinity is the glass phase content. (2) The alkali–acid two-stage dissolution method [16] was used to dissolve the glass phase and f-CaO in the slag. The glass phase content was obtained by subtracting the f-CaO content from the mass reduction rate.

## 3. Results and Discussion

### 3.1. Hydraulic Activity Test

Table 2 shows the hydraulic activity test results of the three kinds of slags, in which  $A_7$  and  $A_{28}$  are the indexes of hydraulic activity for 7 d and 28 d, respectively. It can be seen from Table 2 that the 7 d indexes of hydraulic activity of the two kinds of high-titanium slag are less than 30% of that of blast furnace slag, and the 28 d indexes of hydraulic activity are less than 38% of that of blast furnace slag, so the hydraulic activity of high-titanium slag is far lower than that of blast furnace slag. Comparing the two kinds of high-titanium slag, the index of hydraulic activity for 7 d and 28 d of water-quenched high-titanium slag is 1.7 and 1.5 times that of slow-cooled high-titanium slag, respectively. The hydraulic activity of water-quenched high-titanium slag is obviously higher than that of slow-cooled high-titanium slag.  $K_a$  is the pozzolanic activity rate of slag. The order of the pozzolanic activity rate and the index of hydraulic activity of the three kinds of slag are the same: blast furnace slag > water-quenched high-titanium slag > slow-cooled high-titanium slag. The pozzolanic activity rate of water-quenched high-titanium slag is 57% of that of blast furnace slag, while the index of hydraulic activity of water-quenched high-titanium slag is less than 38% of that of blast furnace slag. Referring to Table 1, it can be seen that the active  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  in high-titanium slag not only have a low dissolution rate, but also have a lower content of  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  than blast furnace slag, due to the high content of  $\text{TiO}_2$ .

**Table 2.** Hydraulic activity test of slag (%).

Sample	A <sub>7</sub>	A <sub>28</sub>	K <sub>4</sub>
PS	125.6	95.6	99.3
TS	35.1	35.8	56.8
TM	21.2	23.5	35.7

3.2. Mineral Phase Analysis

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Figure 1 shows the XRD patterns of three kinds of slag, all of which were thoroughly dried and ground to less than 5 μm. There are broad scattering peaks of glass phase between 21° and 37° in

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the PS and TS patterns that belong to water-quenched slag. The results show that whether it is blast furnace slag or high-titanium slag, if it is water-quenched, the main mineral composition is glass phase. PS and TS patterns that belong to water-quenched slag. The results show that whether it is blast

A small amount of crystalline phases in blast furnace slag, if it is blast water furnace-quenched, slag contains mainly mineral gehlenite composition (2CaO·Al<sub>2</sub>O<sub>3</sub>·SiO<sub>2</sub>) and

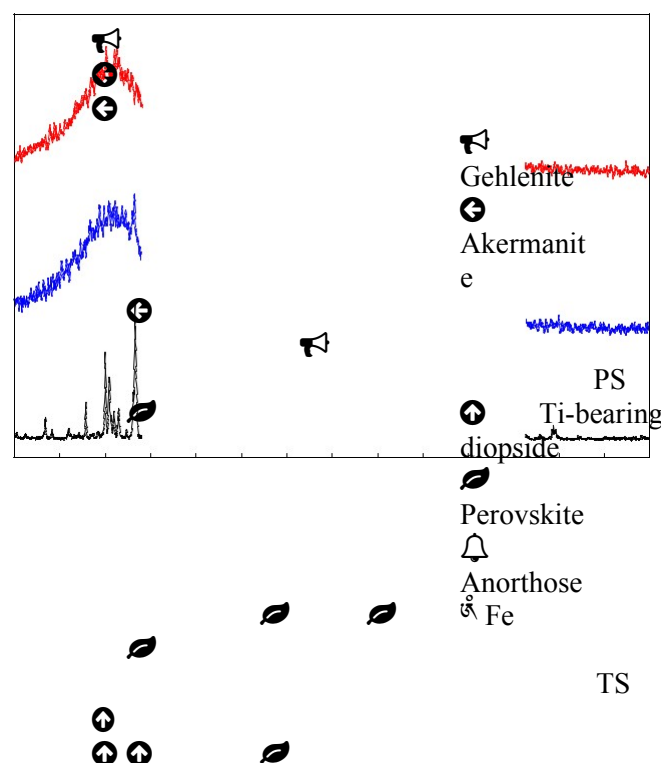
phase. A small amount of crystalline phases in blast furnace slag contain gehlenite akermanite (2CaO·MgO·2SiO<sub>2</sub>); there is no gehlenite or akermanite phase, and only a small amount (2CaO·Al<sub>2</sub>O<sub>3</sub>·SiO<sub>2</sub>) and akermanite (2CaO·MgO·2SiO<sub>2</sub>); there is no gehlenite or akermanite phase, and

of perovskite in water-quenched high-titanium slag. In slow-cooled high-titanium slag, there are only a small amount of perovskite in water-quenched high-titanium slag. In slow-cooled high-titanium slag, there are almost no broad scattering peaks of the glass phase, most of which are crystal minerals. In addition to titanium slag, there are almost no broad scattering peaks of the glass phase, most of which are crystal

perovskite, mineral there is. In also addition Ti-bearing to perovskite, diopside, there is anorthose, Ti-bearing and diopside, iron. anorthose, AsTiO<sub>2</sub> is and iron good. As nucleating TiO<sub>2</sub> is agent,

a good nucleating agent, the high-titanium slag contains perovskite, no matter which cooling method the high-titanium slag contains perovskite, no matter which cooling method we use.

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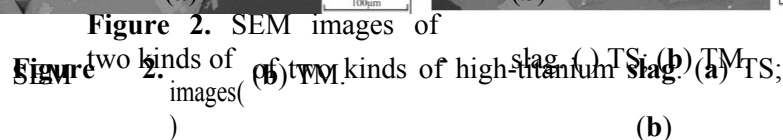






Materials 2020, 13, x FOR PEER REVIEW 12 of 12

embedded in Ti-bearing diopside. In a few bright areas 5, the chemical composition of the micro-region is Fe. It is a residual metallic iron phase in high-titanium slag, with a grain size of approximately 5 to 15  $\mu\text{m}$  and a round or long shape.



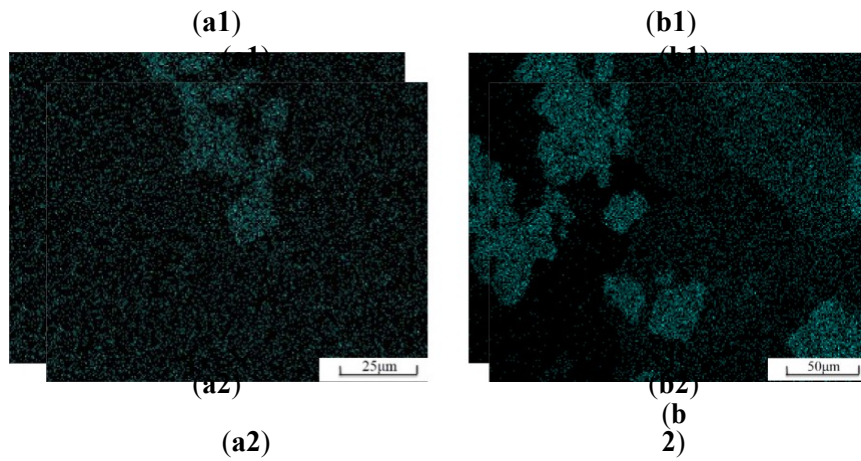
**Figure 2.** SEM images of two kinds of slag. (a) TS; (b) TM.

Figure 3 is a surface image of the Ti distribution in high-titanium slag, where (a1) and (a2)

In water-quenched high-titanium slag, high-Ti is uniformly (b) dispersed throughout the glass phase, and the content of Ti in perovskite is higher than that of glass phase. In slow-cooled

the distribution of Ti is uneven, and the anorthose and iron phases do not contain Ti. The matrix mineral, high-titanium slag, the distribution of Ti is uneven, and the anorthose and iron phases do not contain Ti content.

Figure 10 consists of two backscattered electron (BSE) images of a matrix. Image (a) on the left shows a matrix with Ti content labeled 1 and 2, with a 25µm scale bar. Image (b) on the right shows a matrix with Ti content labeled 1, 2, 3, 4, and 5, with a 50µm scale bar.



**Figure 3.** SEM images of Ti surface scanning of high-titanium slag. **(a<sub>1</sub>)** SEM image of TS; **(a<sub>2</sub>)** Ti surface scanning of TS; **(b<sub>1</sub>)** SEM image of TM; **(b<sub>2</sub>)** Ti surface scanning of TM.



### 3.2.3. Glass Phase Content

In order to ensure the accuracy of the test results, the XRD method and the alkali–acid two-stage dissolution method were used to measure the glass phase of three kinds of slag. The test results are shown in Table 3.

**Table 3.** Comparison of glass phase content in slag (%).

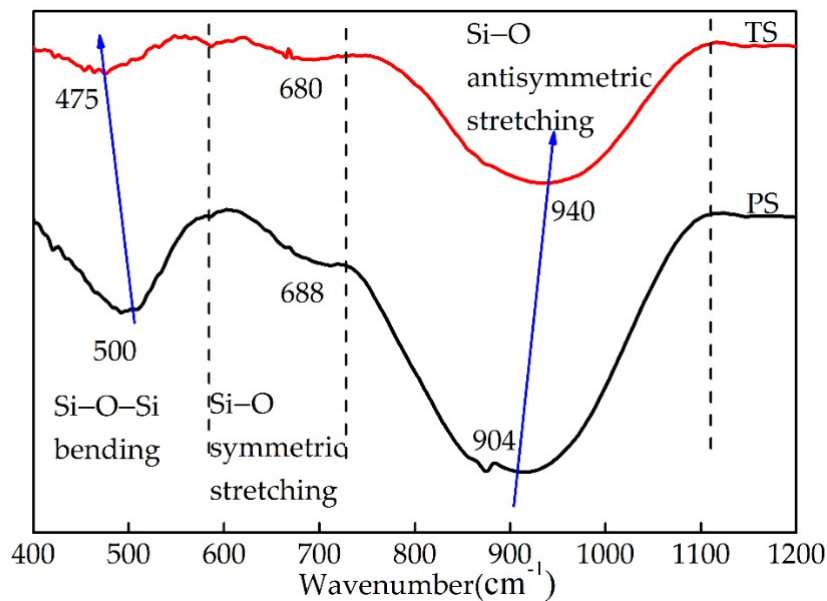
Method	PS	TS	TM
XRD	98.03	97.34	1.82
alkali-acid two stage dissolution	99.49	98.35	1.91

As can be seen from Table 3, the glass phase content of the slow-cooled high-titanium slag is quite low. The glass phase content of the two kinds of water-quenched slag is about 98%, and the results are similar. The presence of more visible perovskites in Figure 2; Figure 3 is shown intentionally to show the morphology of the perovskites, although this does not mean the perovskite content in water-quenched high-titanium slag is high. From this point of view, the hydraulic activity of the two kinds of water-quenched slag is quite different, and it is not caused by the difference in the glass phase content. The microstructure of the two kinds of glass phase should be analyzed.

### 3.3. Analysis of Glass Phase Microstructure

#### 3.3.1. FTIR Spectroscopy Analysis

Figure 4 shows the comparative FTIR spectroscopy of the two kinds of water-quenched slag. It can be seen from the Figure 4 that the effective spectral range of  $400\text{--}1200\text{ cm}^{-1}$  is composed of three regions:  $800\text{--}1000\text{ cm}^{-1}$  with strong intensity,  $400\text{--}600\text{ cm}^{-1}$  with medium intensity, and  $600\text{--}700\text{ cm}^{-1}$  with weak intensity. The  $800\text{--}1000\text{ cm}^{-1}$  band is produced by the antisymmetric stretching vibration of the Si–O bond in the  $\text{SiO}_4$  tetrahedron in slag [17]. Compared with blast furnace slag, the peak position of water-quenched high-titanium slag here is shifted toward a higher wave number. The  $400\text{--}600\text{ cm}^{-1}$  band is produced by the symmetric bending vibration of the Si–O–Si bond. Compared with blast furnace slag, the peak position of water-quenched high-titanium slag here is shifted toward a lower wavenumber. The  $600\text{--}700\text{ cm}^{-1}$  band is produced by the symmetric stretching vibration of the Si–O bond [18]. There was no significant change in the peak position of this band in the two kinds of slag. Sun et al. thought that the  $800\text{--}1000\text{ cm}^{-1}$  band shifted in the high wave number direction, and the  $400\text{--}600\text{ cm}^{-1}$  band shifted toward the low wave number direction, indicating that the degree of polymerization of the glass phase had increased [17]. Therefore, compared with blast furnace slag, the glass phase polymerization degree of water-quenched high-titanium slag increased and the hydraulic activity decreased.



**Figure 4.** FTIR spectroscopy of two kinds of water-quenched slag.

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### 3.3.2. Raman Spectrum Analysis

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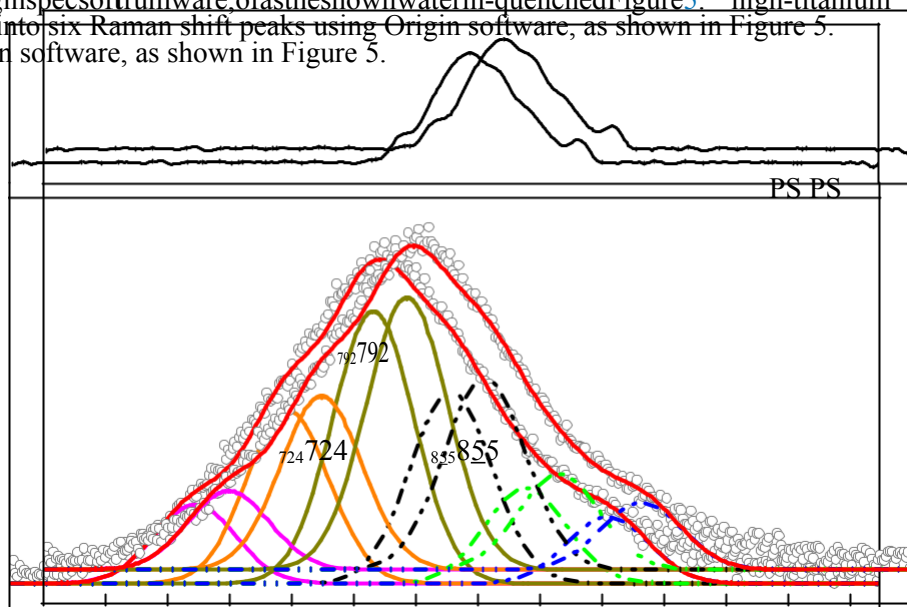
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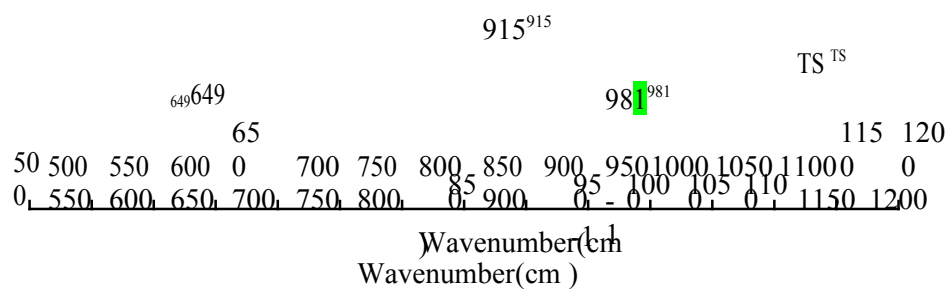
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**Figure 5.** Raman spectroscopy of two kinds of water-quenched slag.  
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In the silicate glass phase system, the structural units of the  $\text{SiO}_4$  tetrahedron can be divided into  $Q^0$ ,  $Q^1$ ,  $Q^2$ ,  $Q^3$ , and  $Q^4$  according to the coordination number of the bridge oxygen around Si. Here,  $n$  in  $Q^n$  represents the coordination number of the bridge oxygen around Si. The lower the bridge oxygen coordination number, the higher the hydraulic activity. The Raman shift for blast furnace slag at  $870\text{ cm}^{-1}$  was produced by the stretching vibration of the Si–O bond in  $Q^0$  [19,20]. The integrated envelopes for water-quenched high-titanium slag were divided into six Raman shift peaks. Of these Raman shift peaks, the Raman shift at  $792\text{ cm}^{-1}$  is generated by the  $\text{Ti–O}^2$  stretching vibration in  $\text{TiO}_4^{4-}$  monomers, the Raman shift at  $724\text{ cm}^{-1}$  is generated by the deformation of O–Ti–O in chain or sheet units [21,22]. The Raman shift at  $649\text{ cm}^{-1}$  is generated by Ti–O stretching vibrations in 6-coordinated  $\text{Ti}^{4+}$  [23]. The above three Raman shifts indicate that the Ti in the water-quenched high-titanium slag mainly exists as  $\text{TiO}_4^{4-}$  monomers, the secondary form is O–Ti–O in chain or sheet units, and a small amount of Ti exists in the 6-coordinated form. The Raman shifts at  $855\text{ cm}^{-1}$ ,  $915\text{ cm}^{-1}$ , and  $981\text{ cm}^{-1}$  are generated by the stretching vibration of the Si–O bond in  $Q^0$ ,  $Q^1$ , and  $Q^2$  [24–26], respectively. The results show  $Q^0$ ,  $Q^1$ , and  $Q^2$  structural units of the  $\text{SiO}_4$  tetrahedron in water-quenched high-titanium slag.

It can be seen from the above test results that the  $\text{SiO}_4$  tetrahedron in blast furnace slag is mainly monosilicate represented by  $Q^0$ , so its hydraulic activity is high.  $\text{TiO}_4^{4-}$  monomers in water-quenched high-titanium slag are more stable than the  $\text{SiO}_4$  tetrahedron [23]. At the same time, there is O–Ti–O in chain or sheet units, which transform the original non-bridged oxygen to bridged oxygen, so that the number of bridged oxygen per  $\text{SiO}_4$  tetrahedron increases, and the degree of network polymerization is enhanced. The Ti in water-quenched high-titanium slag enters the silicate glass system in the form of a network formers, which not only reduces the content of the  $\text{SiO}_4$  tetrahedron, but also transforms some of the  $\text{SiO}_4$  tetrahedron to the more stable form of  $Q^1$  disilicates and  $Q^2$  chain middle groups. Therefore, the glass phase structure of water-quenched high-titanium slag is more stable than that of blast furnace slag, and its hydraulic activity is lower than that.

### 3.3.3. NMR Analysis

#### (1) $^{29}\text{Si}$ spectrum and $^{27}\text{Al}$ chemical environment

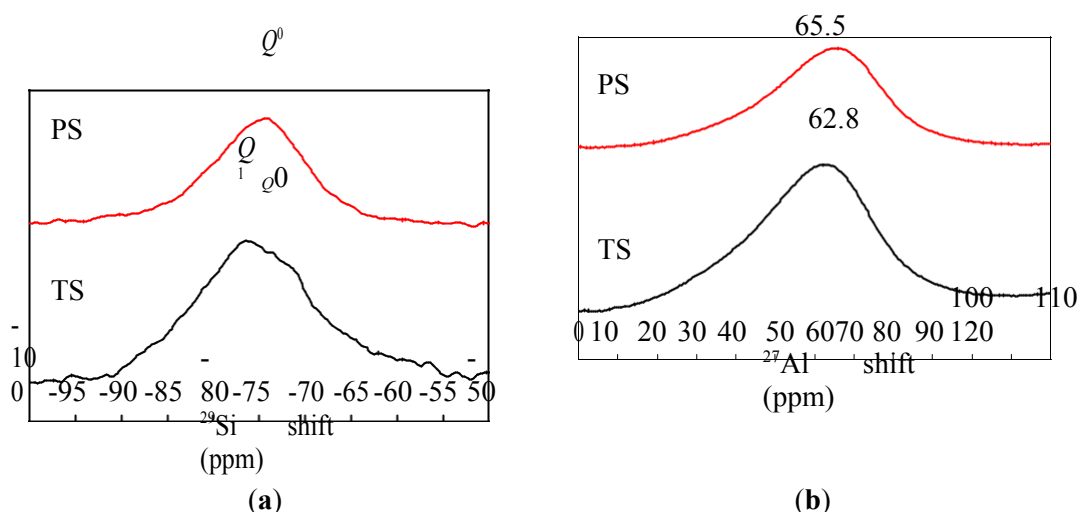
The relationship between chemical shift and structure in  $^{29}\text{Si}$  NMR is shown in Table 4. Al mainly exists in the form of four-coordination or six-coordination in the glass phase network in aluminosilicate. The chemical shift between +50 and +80 ppm belongs to the  $[\text{AlO}_4]$  tetrahedral structure, and the chemical shift between –10 and +20 ppm belongs to the  $[\text{AlO}_6]$  hexahedral structure [27].

**Table 4.**  $^{29}\text{Si}$  NMR chemical shifts of the structure unit of  $Q^n$  in solid silicates [28].

Types of Si-O-X group	Symbol	Chemical Shift/ppm
Monosilicate	$Q^0$	68
		– 76
		76
Disilicates and chain end groups	$Q^1$	– 82
		82
Chain middle groups	$Q^2$	– 88
		88
Layers and chain branching sites	$Q^3$	– 98
Three-dimensional networks	$Q^4$	98–129

Figure 6 shows the  $^{29}\text{Si}$  and  $^{27}\text{Al}$  NMR spectroscopy for two kinds of water-quenched slag, and Figure 6a shows the NMR spectrum of  $^{29}\text{Si}$ . It can be seen from Figure 6a that there is only  $Q^0$  peak in blast furnace slag, indicating that the  $\text{SiO}_4$  tetrahedron in blast furnace slag only exists in the form of monosilicate. Relative to blast furnace slag, the chemical shift of water-quenched high-titanium slag moves to the high field, and the peak is not a  $Q^0$  single peak, but the superposition peak of  $Q^0$  and  $Q^1$ . This shows that the  $\text{SiO}_4$  tetrahedron in water-quenched high-titanium slag exists in the form of monosilicate and disilicates, which is consistent with the analysis of FTIR spectroscopy and Raman spectroscopy. Figure 6b shows the NMR spectrum of  $^{27}\text{Al}$ . The  $^{27}\text{Al}$  peaks of blast furnace slag and water-quenched high-titanium slag are 65.5 ppm and 62.8 ppm, respectively, so the chemical

environment of Al element in the two kinds of slag is  $[\text{AlO}_4]$  tetrahedral, and the presence of Ti has no obvious influence. *Materials* **2020**, *13*, xFOR on PE ER the structure REVIEW of Al.



**Figure 66.**  $^{29}\text{Si}$  and  $^{27}\text{Al}$  NMR spectroscopy for two kinds of water-quenched slag. (a)  $^{29}\text{Si}$  NMR spectroscopy; (b)  $^{27}\text{Al}$  NMR spectroscopy.

(2) Characterization of polymerization degree

The polymerization degree is an index reflecting the comprehensive influence of the chemical composition and glass phase structure on the hydraulic activity. The higher the polymerization degree, and glass phase structure on the hydraulic activity. The higher the polymerization degree, the lower the hydraulic activity. The change in the oxygen number can be used to reflect degree, the lower the hydraulic activity. The change in the bridging oxygen number can be used to reflect the polymerization degree of the system. The bridging oxygen number increases with Si–O bond reflect the polymerization degree of the system. The bridging oxygen number increases with Si–O polymerization and decreases with Si–O–Si bond depolymerization. Therefore, the higher the bridging bond polymerization and decreases with Si–O–Si bond depolymerization. Therefore, the higher the oxygen number, the higher the degree of polymerization and the lower the hydraulic activity. bridging oxygen number, the higher the degree of polymerization and the lower the hydraulic activity. In this study, the deconvolution of the  $^{29}\text{Si}$  NMR spectroscopy of TS was performed using activity.

Origin software, and the relative areas of the resonance peaks were calculated. The magnitude of

In this study, the deconvolution of the  $^{29}\text{Si}$  NMR spectroscopy of TS was performed using Origin software, and the relative areas of the resonance peaks were calculated. The magnitude of

corresponding relative bridging oxygen (RBO) number was calculated according to the following formula [29]:

$$\text{RBO} = \frac{1}{4} \left( \frac{Q^1}{Q^1 + Q^2 + Q^3 + Q^4} + 2 \frac{Q^2}{Q^1 + Q^2 + Q^3 + Q^4} + 3 \frac{Q^3}{Q^1 + Q^2 + Q^3 + Q^4} + 4 \frac{Q^4}{Q^1 + Q^2 + Q^3 + Q^4} \right) \quad (1)$$

$$\text{RBO} = \frac{1 \times \text{area of } Q^1 + 2 \times \text{area of } Q^2 + 3 \times \text{area of } Q^3 + 4 \times \text{area of } Q^4}{\sum \text{area of } Q^i} \quad (1)$$

where  $Q^i$  is the relative area of the corresponding resonance peak.

As shown in Figure 7, the  $^{29}\text{Si}$  NMR spectroscopy of water-quenched high-titanium slag is deconvoluted into five peaks. Based on classification according to the chemical shift, there are two

As shown in Figure 7, the  $^{29}\text{Si}$  NMR spectroscopy of water-quenched high-titanium slag is deconvoluted into five peaks. Based on classification according to the chemical shift, there are two  $Q^1$  peaks, two  $Q^0$  peaks, and one  $Q^2$  peak. Therefore, the  $Q^1$ ,  $Q^0$ ,  $Q^2$  forms of the  $\text{SiO}_4$  tetrahedron

exist in the water-quenched high-titanium slag. Because  $Q^1$  has the largest area, it can be said that peaks, two  $Q^0$  peaks, and one  $Q^2$  peak. Therefore, the  $Q^1$ ,  $Q^0$ ,  $Q^2$  forms of the  $SiO_4$  tetrahedron exist. Disilicate is the main form in water-quenched high-titanium slag. After calculating the integral area of in the water-quenched high-titanium slag. Because  $Q^1$  has the largest area, it can be said that disilicate

each peak, the relative areas of  $Q^1$ ,  $Q^0$ , and  $Q^2$  were found to be 100, 81.06, and 21.82, respectively, is the main form in water-quenched high-titanium slag. After calculating the integral area of each and the calculated RBO value was 0.2. Blast furnace slag has only one  $Q^0$  single peak, consequently, its peak, the relative areas of  $Q^1$ ,  $Q^0$ , and  $Q^2$  were found to be 100, 81.06, and 21.82, respectively, and the

RBO was 0. Therefore, the high degree of polymerization and more stable structure of the glass phase calculated RBO value was 0.2. Blast furnace slag has only one  $Q^0$  single peak, consequently, its RBO in water-quenched high-titanium slag are the main reasons for its low hydraulic activity.

was 0. Therefore, the high degree of polymerization and more stable structure of the glass phase in water-quenched high-titanium slag are the main reasons for its low hydraulic activity.



## 4. Conclusions

### 4. Conclusion

The main mineral phase composition of the slow-cooled high-titanium slag included perovskite, Ti-bearing diopside, and anorthose. Because these mineral phases pertain to inert mineral crystals, and the content of glass phase was less than 2%, the hydration activity was low. Although the content

and the content of glass phase was less than 2%, the hydration activity was low. Although the content

of glass phase in water-quenched high-titanium slag was 98%, Ti mainly existed in the glass phase of glass phase in water-quenched high-titanium slag was 98%, Ti mainly existed in the glass phase structure in the form of  $\text{TiO}_4$  monomers, chain or sheet units, and a small amount of 6-coordination,

structure in the form of  $\text{TiO}_4$  monomers, chain or sheet units, and a small amount of 6-coordination, which were relatively stable. The titanium oxygen structure in the glass phase not only reduced the

which were relatively stable. The titanium oxygen structure in the glass phase not only reduced the relative content of  $\text{SiO}_4$  tetrahedron, but also made the  $\text{SiO}_4$  tetrahedron in the glass phase network a relative content of  $\text{SiO}$  tetrahedron, but also made the  $\text{SiO}$  tetrahedron in the glass phase network a monosilicate, and more stable forms of disilicates and chain middle groups appeared. The relative monosilicate, and more stable forms of disilicates and chain middle groups appeared. The relative bridge oxygen number increased to 0.2, so the hydraulic activity of the water-quenched high-titanium

bridge oxygen number increased to 0.2, so the hydraulic activity of the water-quenched high-titanium slag was also low. Based on the results of this study, the removal of Ti from high-titanium slag not only increased the relative content of active  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  in the glass phase, but also improved the only increased the relative content of active  $\text{SiO}_2$  and  $\text{Al}_2\text{O}_3$  in the glass phase, but also improved the hydraulic activity of the glass phase.

hydraulic activity of the glass phase.

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### Nomenclature

XRD Nomenclature X-ray Di

raction

SEM Scanning Electron Microscope

XRD X-ray Diffraction

EDS Energy Dispersive Spectrometer

SEM Scanning Electron Microscope

BEI Back Scattering Electron Image

FTIR Fourier Transform Infrared spectroscopy

EDS Energy Dispersive Spectrometer

NMR Nuclear Magnetic Resonance spectroscopy

BEI Back Scattering Electron Image

TS Water-quenched high-titanium slag

FTIR Fourier Transform Infrared spectroscopy

PS Blast furnace slag

TM NMR Slow-cooling Nuclear Magnetic Resonance spectroscopy

A Index of hydraulic activity for 7 d slag

T Water-quenched high-titanium

S7

PS Blast furnace slag

TM Slow-cooling high-titanium slag

$K_a^{28}$	Index of hydraulic activity for 28 d
$Q^n$	The pozzolanic activity rate
RB	Types of Si–O–X group, n is the bridge oxygen
O	number of coordination around Si
	Relative bridge oxygen number

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