

Article

Rejuvenation of Zr-Based Bulk Metallic Glasses by Ultrasonic Vibration-Assisted Elastic Deformation

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Abstract: The rejuvenation of Zr_{52.5}Cu_{17.9}Ni_{14.6}Al₁₀Ti₅ bulk metallic glasses (BMGs) by ultrasonic vibration-assisted elastic deformation (UVEF) was studied herein. The UVEF-treated samples demonstrate an obvious rejuvenation and have a higher relaxation enthalpy and a smaller range of supercooled liquid regions than the as-cast samples. The fracture of the rejuvenated amorphous alloy is mainly ductile fracture, and shear deformation occurs in the deformation region. It is also found that as the amplitude increases, the free volume of the rejuvenated amorphous alloy increases, the yield strength and the elastic modulus decrease, and the formability increases. The free-volume content is used to characterize the degree of rejuvenation, and a mathematical model of the relationship between the ultrasonic amplitude and free volume is established. In addition, it is found that the ultrasonic vibration stress induces the additional free volume in the Zr-based bulk metallic glasses and improves the plasticizing behavior. The temperature rise caused by the ultrasonic thermal e ect does not induce additional free volume.

Keywords: bulk metallic glasses (BMGs); rejuvenation; ultrasonic vibration-assisted elastic deformation (UVEF); free volume; plasticity

1. Introduction

Bulk metallic glasses (BMGs) are materials with excellent performance and mechanical properties, such as high yield strength [1], high toughness [2], high hardness [3] and exceptional "damage tolerance" [4,5]. However, the nonuniform deformation caused by the highly localized deformation at room temperature directly causes its brittleness [6,7], which has become the bottleneck for the widespread application of BMGs. Reducing the brittleness of BMGs has become a scientific issue that scholars have begun to pay attention to.

The rejuvenation of amorphous alloys occurs when BMGs are injected with high energy because it endows them with additional free volume and greater plasticity [8,9]. Therefore, rejuvenation of BMGs to reduce their brittleness and improve their plasticity at room temperature has attracted a lot of attention [10–12]. For example, J. Michler [13] conducted compression experiments on ion-radiated BMGs and found that they can increase the free-volume content of BMGs and enhance their plasticity. J. Das [14] utilized cold-rolled Zr-based BMGs at room temperature to overcome their inherent brittleness and make BMGs plastic from 0.5% to 15%. F. X. Li [15] discovered that during the elastic compression of

Zr₃₅Ti₃₀Be_{27.5}Cu_{7.5} BMGs at room temperature, the BMGs transitioned from mechanical relaxation to rejuvenation, thereby improving their plasticity. W. Guo [16] studied the rejuvenation of Zr–Cu–Al–Ni–Ta BMGs under cryogenic cycle treatment (DCT) and found that by adding more Ta, the BMGs can be restored to an elevated energy, and the compressive fracture strength and plastic strain increased with increasing Ta content. Through plastic deformation under triaxial compression at room temperature,

Y. Li and A.L. Greer [4] rejuvenated BMGs samples and found that the rejuvenated amorphous material showed excellent plastic deformation ability. However, elastic loading [17], ion radiation [18], thermal cycling [19,20], and plastic deformation [21] methods require a substantial amount of time to rejuvenate the BMGs, and during this time, a relaxation effect inevitably occurs, weakening the rejuvenation effect.

In addition, ultrasonic vibration-assisted has an ultrasonic stress e ect and ultrasonic thermal e ect. P. Chen [22] used ultrasonic vibration to process a Ti-based amorphous powder to make nanocrystalline bulk materials. The amorphous powder was crystallized by the frictional heat generated during the ultrasonic vibration and then welded to form a bulk nanocrystalline material. J. Ma et al. [23] used supersonic vibration to stamp BMGs and found that this forming method completed thermoplastic forming in seconds and largely avoided the time-dependent crystallization and oxidation processes, thus avoiding the risk of crystallization that occurs during traditional heat processing. N. Li et al. [24] conducted uniaxial tensile and compression experiments on Zr-based amorphous alloys under the action of a vibration field and found that with increasing vibration frequency, the free-volume content of the amorphous alloy increased and the flow unit volume decreased, which caused the flow viscosity to decrease and the micro-forming ability to increase. However, there are no reports that specifically analyze the influence of ultrasonic vibration stresses and ultrasonic thermal e ects on amorphous properties.

A novel method of ultrasonic vibration-assisted elastic deformation (UVEF) was developed that can successfully and rapidly rejuvenate Zr-based BMGs within 8 s. It was found that ultrasound-assisted vibration can rapidly increase the internal energy of BMGs and quickly drive the loosely packed atoms in BMGs to high-energy regions, thereby driving additional areas with free volume and rheological units to form shear bands. These processes result in the rapid rejuvenation of BMGs and prevent the time-dependent crystallization and relaxation phenomena. In addition, the thermodynamics, mechanical properties and fracture morphology of the rejuvenated UVEF-treated samples were analyzed. It was found that as the amplitude increased, the yield strength and elastic modulus of the UVEF-treated samples decreased, and the plasticity was greatly improved. The free volume was used to characterize the degree of rejuvenation of the amorphous alloys herein, and a mathematical model of the relationship between the ultrasonic amplitude and the free volume of the rejuvenated BMGs was established. The e ects of the temperature rise and stress caused by the ultrasonic vibration on the rejuvenation properties of BMGs were analyzed. These findings provide a convenient and fast method to reduce the room-temperature brittleness of BMGs and improve their plasticity.

2. Experimental Methods

2.1. Sample Preparation

The experimental material is an amorphous alloy with an alloy composition of Zr_{52.5}Cu_{17.9}Ni_{14.6}Al₁₀Ti₅ (atomic percentage). The raw materials of metals Zr, Cu, Ni, Al, and Ti with a purity of 99.9% or more are prepared according to the nominal composition. An electric arc furnace is used to melt the master alloy, and then a copper die suction casting method is used to prepare rod-shaped amorphous particles with a diameter of F 2 mm. A low-speed diamond cutting machine is used to process a cylindrical sample with dimensions of F 2 mm 4 mm. The ends of the cylinder are also polished to ensure that both ends are parallel to each other and orthogonal to the centerline of the cylinder. The sample was analyzed with X-ray di raction (XRD, Bruker D8 Advance, Bruker, Karlsruhe, Germany) to ensure that the sample is in an amorphous state.

2.2. UVEF Processing

The experiments herein use a homemade ultrasonic vibration-assisted compression test platform. The ultrasonic vibration-assisted device is installed on a Zwick Z050 tensile testing machine (Zwick Roell Group, Ulm, Germany). The maximum power of the ultrasonic system is 1500 W, and the vibration frequency is 20 kHz. Due to the limitation of the experimental conditions, the ultrasonic frequency in

 $this 36, and study 43\ is \mu muno. The anged; experiment only the is\ amplitude divided into is\ change two steps d, and First, the the amplitude as amplitude are positioned 19,27,36, and and clamped 43.$ Thebyaexperimentfixture.Next,is dividedanultrasonicintotwovibrationsteps. isFirst,appliedthe sampletothesampleispositionedwhiletheandfixtureclampedandbythea samplefixture Narext,fixedan (Figureultrasonic1). Thevibrationspeedisis appliedV=V0+ toV ulthe(t),samplewhere Vwhileoisthethepressingfixture speed,andtheV ulsampleistheultrasonicarefixed (Figurevibration1). speed, The speed and ist is V = the Votime + V.ulA(t), strain wherate Voof is the pressing 0.01-1 and speed, a strain Vulis of the ultrasonic 2 were vibration used to speed, ensure and compression tist he time within. A strain the amorphousrateof" = ¹deformationandstrain rangeof"=. In2%addition,wereusedduringtoensurethe compressionultrasonicyibrationwithin-assistedtheamorphouselasticcompressionelasticdeformationdeformation, ragea. highlnaddition,-frequencyduringvibrationthe ultrasoniccausesthe vibrationsamplesto-assistedslip, soelastic preload compressiforce of n50 deform Nisapplied tion, highfirst-tofrequencyfixthesamples,vibrationandcausthensthesamplesultrasonicto slvibrationp,sopreloadisappliedforceuntilof 50theNelasticisappliedcompressionfirsttofixdeformationthesamples,ends,and asthenshowntheultrasonicinFigure vibration1.Thetimels appliedrequireduntilforUVEFtheelasticcanbecompressioncalculated: deformation ends, as shown in Figure 1. The time required for

UVEF can be calculated:

where t_{pl} is the time of of preloading, t_{pl} is the time of of preloading, t_{pl} is the initial punch and and the upper t_{pl} surface of the of sample, the sample, and t_{pl} is the preloading time, t_{pl} is the preloading time, t_{pl} is the preloading t_{pl} is th $^{\prime}$ and $^{\prime}$ and $^{\prime}$ and a strain of " = 2% were were taken as 0.5 mm and 0.1 mm/s, respectively. A strain rate of " = 0.01 s 1 and a strain of " = 2% were were taken as 0.5 mm and 0.1 mm/s, respectively. A strain rate of 0.01 s-1 and a strain of 2% $applied_{were applied} to ensure_{to}$

 $ensure that compression that compression is within is the within amorphous the amorphous elastic deformation elastic deformation range, BMGs\ and can be a surface of the compression of the compression$ $\texttt{be}_{BMGs} \textit{rejuve}_{can} \\ \texttt{at}_{be} \\ \texttt{d}_{rejuvenated} \\ \textit{within} \\ \texttt{BMGs}_{8s.} \\ \textit{samples} \\ \textit{Five}_{within} \\ \texttt{BMGs}_{8s.} \\ \textit{samples} \\ \textit{Five}_{bMGs} \\ \textit{were}_{samples} \\ \textit{reeatedly}_{were} \\ \textit{compressed}_{repeatedly} \\ \textit{descending}_{samples} \\ \textit{descen$ ${\sf under}_{compressed} {\sf achdeformation}_{undereach} \stackrel{{\sf process}}{=} {\sf deformation} \stackrel{{\sf condition}}{=} {\sf process} \stackrel{{\sf condition}}{=} {\sf condition}.$

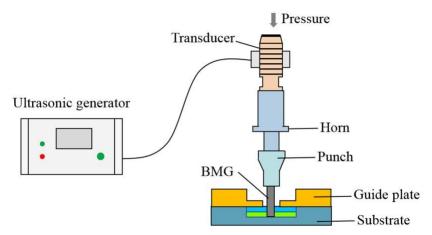


Figure 1.. Ultrasonic vibration--assisted elastic deformation diagram.

2.3. Hot-Compressed Elastic Deformation (HEF) Experiment

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To specialized analyze the e ects of the temperature rise and stress produced during the ultrasonic To specialized analyze the effects

of the temperature rise and stress produced during the vibration on the rejuvenation properties of the amorphous alloys, we eliminate the e ect of ultrasonic ultrasonic vibration on the rejuvenation properties of the amorphous alloys, we eliminate the effect stress, and only the e ect of temperature rise is studied separately. The temperature rise at di erent of ultrasonic stress, and only the effect of temperature rise amplitudes is obtained according to previous research results [25]. That is, amplitudes of 19, 27, 36, at different amplitudes is obtained according to previous research results [25]. That is, amplitudes of and 43 m correspond to temperature increases of 80, 150, 200, and 270 °C. HEF experiments at the 19, 27, 36, and 43 µm correspond to temperature increases of 80, 150, 200, and 270 °C. HEF corresponding temperatures are performed on the thermal simulation machine (Gleeble3800, DSI, experiments at the corresponding temperatures are performed on the thermal simulation machine

Austin, TX, USA). The corresponding strain rate and strain are consistent with the UVEF experiment (Gleeble3800, DSI, Austin, TX, USA). The corresponding strain rate and strain are consistent with the and are 0.01 s⁻¹ and 2%, respectively. Five BMGs samples are repeatedly hot-compressed under each UVEF experiment and are 0.01 s⁻¹ and 2%, respectively. Five BMGs samples are repeatedly hot-compressed under each under

compressed under each deformation condition.

2.4. Analytical Testing

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To study the properties of the Zr-based BMGs samples, a series of analytical tests are performed on To study the properties of the Zr-based BMGs samples, a series of analytical tests are performed them. XRD (Bruker D8 Advance, Bruker, Karlsruhe, Germany) is used to determine the crystallization on them. XRD (Bruker D8 Advance, Bruker, Karlsruhe, Germany) is used to determine the of the samples. The scanning range is 20 ~80 , the step size is 0.02 step 1, and the scanning speed is crystallization of the samples. The scanning range is 20° ~ 80°, the step size is 0.02° step-1, and the 12 min 1. The Vickers microhardness test (BUEHLER 5103, Buehler, Lake Blu , IL, USA; ASTM 92-82, scanning speed is 12°-min-1. The Vickers microhardness test (BUEHLER 5103, Buehler, Lake Bluff, IL,

USA; ASTM E92-82, West Conshohocken, PA, USA) is used to determine the hardness of the samples, the diamond indenter was pressed into the surface of the specimen along its axial direction with a

force of 50 gf, and then the force was held for 10 s. Differential scanning calorimetry (DSC,

PerkinElmer focewasheld for 10 s. Di erential scanning calorimetry (DSC, PerkinElmer DSC8000, hermodynamic PerkinEler, DSC8000, PerkinElmer, Poulsbo, WA, USA) is used to analyze the

Poulsbo, WA, USA) is used to analyze the hermodynamic characteristics of the samples at a heating

3. Results and Discussion

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Figure 2 is a comparison of the XRD patterns of the as-cast and UVEF-treated samples. It can Figure 2 is a comparison of the XRD patterns of the as-cast and UVEF-treated samples. It can be be found that the XRD diffraction peaks of the UVEF-treated samples are consistent with those of found that the XRD diffraction peaks of the UVEF-treated samples are consistent with those of the the as-cast samples. There are broad diffusion peaks without any obvious crystalline peaks in both as-cast samples. There are broad diffusion peaks without any obvious crystalline peaks in both cases, cases, which indicates that the as-cast and UVEF-treated samples are all amorphous, and ultrasonic which indicates that the as-cast and UVEF-treated samples are all amorphous, and ultrasonic vibration-assisted elastic deformation does not cause the amorphous material to crystallize. Regarding the vibration-assisted elastic deformation does not cause the amorphous material to crystallize. naming convention, the UVEF-19 sample is an amorphous sample treated by ultrasonic vibration-assisted Regarding the naming convention, the UVEF-19 sample is an amorphous sample treated by elastic deformation with an amplitude of 19 m.

ultrasonic vibration-assisted elastic deformation with an amplitude of 19 µm.

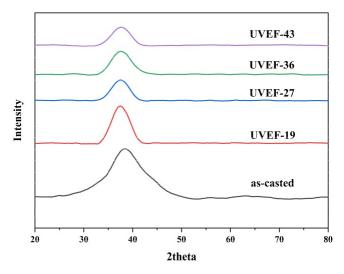
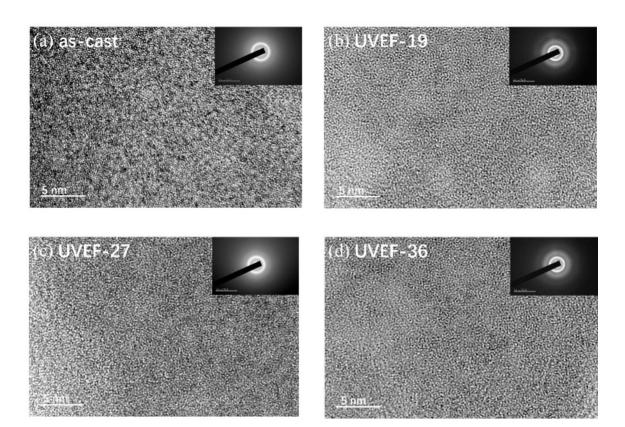


Figure 2.. Comparison of the XX-ray-ray diffraction (XRD) (XRD) patterns patterns of the of as the -castas-- and castultrasonic and ultrasonic vibration vibrati

assisted-assisted elasticde formation (UVEF) -treated (UVEF) samples.

To furtherconfirmconfirmthetheamorphous properties of the asthe-cast-andcast UVEFand-UVEFtreated-treated samples, samples, high-resolution high-resolution TEM is TEMused is to used obtain to an obtain image ano finage them icrostructure of the microstructure of the sample, of the which sample, is shown which in its Figure shown 3 in. No Figure nanocrystals 3. Nonanocrystals are found are in the foundas-cast inhe samples as-cast or samples the UVEF or the-treated UVEF sample-treates, amples, and the microstructure and the microstructure of all samples of all shows samples labyrinths hows-like labyrinthcharacteristics-like characteristics similar to those similar of anto amorphous which state, is consistent which is with consistent the XRD with test the results.



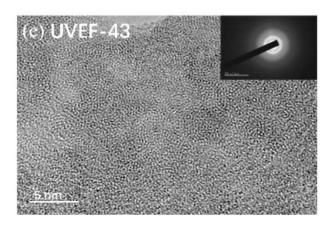


Figure 3. Transmission electron microscopy (TEM)) images of the as-cast sample and the UVEF-treated Figure 3. Transmission electron microscopy (TEM)) images of the as-cast sample and the UVEF- samples: (a) as-cast; (b) UVEF-19; (c) UVEF-27; (d) UVEF-36; and (e) UVEF-43. treated samples: (a) as-cast; (b) UVEF-19; (c) UVEF-27; (d) UVEF-36; and (e) UVEF-43.

3.1. Thermodynamics Analysis

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Thermodynamics Analysis
 The as-cast and UVEF-treated samples are analyzed by DSC, and the thermodynamic properties The as-cast and UVEF-treated samples are analyzed by DSC, and the thermodynamic properties of the samples are obtained from the DSC curve. As shown in Figure 4, the DSC curves of all samples

samples of the samples are obtained from the DSC curve. As shown in Figure 4, the DSC curves of all samples show an endothermic phenomenon starting from the T_g temperature, which is a characteristic of glass show an endothermic phenomenon starting from the T_g temperature, which is a characteristic of glass transition, and then exhibit an exothermic phenomenon corresponding to the crystallization behavior transition, and then exhibit an exothermic phenomenon corresponding to the crystallization behavior to the T_g temperature. The values of T_g and T_g for as-cast and UVEF treated samples as shown in to the T_g temperature. The values of T_g and T_g for as-cast and UVEF treated samples as shown in T_g and T_g for as-cast and T_g for as-

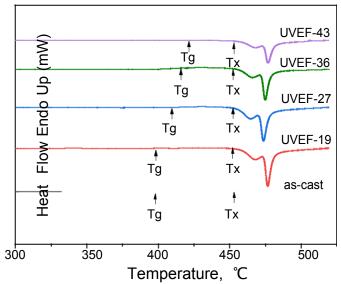


Figure 4. Differential scanning calorimetry (DSC) curves of the as-cast and UVEF-treated samples. Figure 4. Di erential scanning calorimetry (DSC) curves of the as-cast and UVEF-treated samples.

TableTable1. 1The.Thevaluesvaluesof of the the glass glass transition temperature TgTandgand the the crystallization temperature TxTforx for a sascast-castandandUVEFtreatedsamples..

 Sample	T_g (K) T_g (K)	T_X (K) T_x (K)	DT Δ (K) T(K)
As-cast	397 ³⁹⁷	452 ⁴⁵²	55
UVEF-19	398 ³⁹⁸	452 452	₅₄ 54
UVEF-27	410 ⁴¹⁰	453 ⁴⁵³	43
UVEFUVEF-36-36	416 416	453 453	37 37
UVEF-43-43	422 422	454 454	32 32

Compared with that of the as-cast sample, the crystallization temperature Tx of the UVEF-treated Compared with that of the as-cast sample, the crystallization temperature Tx of the UVEF-treated sample shows only slight fluctuations, roughly in the range of 453 ± 1 °C, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations, roughly in the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations on the range of 453 ± 1.0, while the glass transition sample shows only slight fluctuations.

in the range of 453 1 C, while the glass transition' temperature T_s showed an upward trend with increasing ultrasonic amplitude. This shows that an temperature T_S showed an upward trend with increasing ultrasonic amplitude with increasing ultrasonic amplitude ultrasonic amplitude ultrasonic amplitude ultrasonic importance and increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition temperature T_S of the amplitude ultrasonic vibration can increase the glass transition the superature vibration can increase the glass transition that the cannot provide the cannot provide

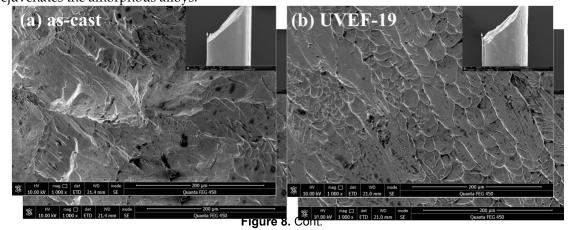
of the UVEF-treated sample increases.

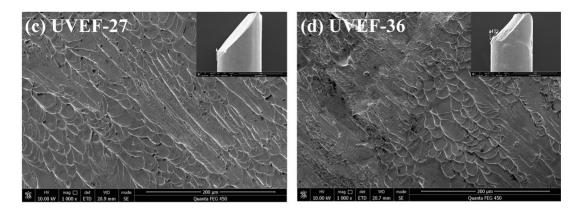




samplesples obtained by by SEM. All Allsamples fracture along along the themaximum shears hears tress stress plane plane at an angle angle of approximately of 45° 45 to to the the loading loading axis axis. The Themicrostructure was was not not uniform across the fracturere surface, if , and the fracturesr off as--cast samples are obviously in the shape of veins, mountains, s, and treatment of the loading as the maximum shear stress plane at an angle angle of approximately of 45° 45 to to the the loading axis axis. The Themicrostructure was was not not uniform across the fracturere surface, if , and the fracturesr off as--cast samples are obviously in the shape of veins, mountains, s, and treatment of the loading as the maximum shear stress plane at an angle angle of approximately of the loading as the maximum shear stress plane at an angle angle of approximately of the loading axis axis. The Themicrostructure was was not not uniform across the tracture of the loading as the maximum shear stress plane at an angle angle of approximately of the surface, and the fracture along the nations, so and or the loading as the maximum shear stress plane at an angle angle of approximately of the surface, and the fracture along the nations plane at an angle angle of approximately of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface, and the surface along the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface and the surface plane at an angle angle of the surface, and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface and the surface plane at an angle angle of the surface plane at an angle angle of the surface plane at

results in Figure 6a, indicating that the ultrasonic vibration-assisted elastic deformation process rejuvenates the amorphous alloys.





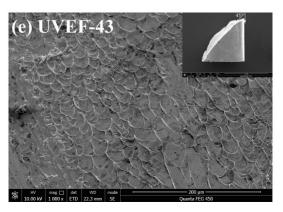


Figure 8. Compressive fracture micromorphology of the as-cast and UVEF-treated samples: (a) as-cast; Figure 8. Compressive fracture micromorphology of the as-cast and UVEF-treated samples: (a) as- (b) UVEF-19; (c) UVEF-27; (d) UVEF-36; and (e) UVEF-43.
cast; (b) UVEF-19; (c) UVEF-27; (d) UVEF-36; and (e) UVEF-43.

3.5. Properties of the Hot-Compressed Elastic Deformation Treated Sample

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From a comparison of XRD patterns of hot-compressed elastic deformation (HEF)-treated samples in FigureFrom9a,a itcomparisoncanbefound of thatXRDthepatternsHEF-treatedofhotsamples-compressedhavebroadelastical deformationusionpeaks and (HEF)no-obvioustreated samplescrystallinein peaks, Figure which9a, itcanprovesbefoundthatthethatHEFthe-treatedHEF-treatedsamplessamplesarestillhaveall amorphousbroaddiffusion. Regardingpeaks and the nosampleobviousnamingcrystallineconvention, peaks, HEFwhich-80 refers provesto amorphous that the HEFsamples-treated processed samples by are hotstill-compressed all amorphous elastic.

Regarding deformation the atsample temperaturenaming of convention, 80 C. The HEFmicroscopic-80 refers morphology to amorphous observed samples with processed TEMshown by hotin-compressed Figure 9 bindicates elastic deformation that the HEF-treated sample shows a maze-like amorphous structure, Figure and 10 anoshows nanocrystal sthemicrobardness are found values. of the as-cast and HEF-treated samples. The microbardness

value does not decrease with increasing temperature. Compared with the UVEF-treated samples (Figure 7), the e ect of deformation on the amorphous particles is the opposite, indicating that only an elastic deformation treatment with increasing temperature does not induce an acrease in the amount of free volume. It is also found that the relaxation enthalpy DH of the HEF-treated sample (Figure 100), but as the temperature increases, the relaxation enthalpy DH of the HEF-treated sample essentially does not change indicating that its free volume does not change with increasing temperature it also further shows that the aspect of the ultrasonic vibration-assisted elastic deformation process that creates additional free volume is the ultrasonic vibration stress can create additional free volums and improve the repvenation degree of amorphous alloys [31]. 2theta

Figure 9. (a) Comparison of XRD patterns of as-cast and hot-compressed elastic deformation (HEF) - treated samples, and (b) TEM image of HEF-270 treated sample.

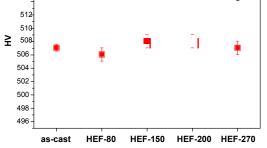
Figure 10a shows the microhardness values of the as-cast and HEF-treated samples. The microhardness value does not decrease with increasing temperature. Compared with the UVEF-treated samples (Figure 7), the effect of deformation on the amorphous particles is the opposite,

samples in Figure 9a, it can be found that the HEF-treated samples have broad diffusion peaks and no obvious crystalline peaks, which proves that the HEF-treated samples are still all amorphous. Regarding the sample naming convention, HEF-80 refers to amorphous samples processed by hotcompressed elastic deformation at a temperature of 80 °C. The microscopic morphology observed MaterialswithTEM2020shown,13,4397in Figure 9b indicates that the HEF-treated sample shows a mazelike amorphous 11 of 15 structure, and no nanocrystals are found.

(a) Materials 2020, 13, x FOR PEER REVIEW HEF-270 11 of 15 increase in the amount of free two times. It is also than the relaxation entirily all of the HEF-treated sample is slightly larger than that of the sound that the relaxation enthalpy art of the HEF-ascest sample (Figure 10b), but as the temperature extend sample essentially does not change sindicating temperature if also turner shows that the aspect of reprocess that creates additional free volume is the rise caused by the ultrasonic thermal affect. The increases, the relaxation enthalog AH of the HEF that its free volume does not change with increasi the ultrasonic vibration-assisted elastic deformat ultrasonic vibration stress, not the temperature 500 me and impro ultrasoftic vibration stress can create additional from rejuvenationFigure9.(a)degreeComparisonofamorphousofXRDpatternsalloy Comparison of XRD patterns of as-cast and hot-compressed elastic deformation (HEF) treated samples, and (b) TEM image of HEF-270 treated sample. -treated samples, and (b) TEM image of HEF-270 treated sample.

Figure 10a shows the(a)microhardness₅₂₀ values of the as-cast and HEF-treated samples. The microhardness value does not 518 decrease with increasing temperature. Compared with the UVEF-

treated samples (Figure 7), the 516,14 effect of deformation on the amorphous particles is the opposite,



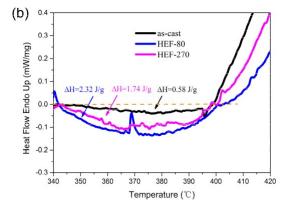


Figure 10. (a) Comparison of microhardness of the as-cast and HEF-treated samples and (b) comparison Figure 10. (a) Comparison of microhardness of the as-cast and HEF-treated samples and (b) of relaxationcomparisonenthalpiesofrelaxationoftheenthalpiesas-castandof the HEF as-treated-castandsamples. HEF-treated samples.

the ultrasonic thermal e ect in this study is 270 C, which is far lower than the T_g of the Zr-based ultrasonic thermal effect in this study is 270 °C, which is far lower than the T_g of the Zr-based BMGs.

It is found that the fracture stress—strain curve of the HEF-treated samples has a very short yield. It is found that the fracture stress—strain curve of the HEF-treated samples has a very short yield. Stage before fracture, approximately 0.3% of the strain, and then the samples immediately fracture stage before fracture, approximately 0.3% of the strain, and then the samples immediately fracture. and then the samples immediately fracture (Figure 11). At the same time, it is found that the yield strength, elastic modulus, and plastic strain of (Figure 11). At the same time, it is found that the yield strength, elastic modulus, and plastic strain of the HEF-treated sample essentially did not change with increasing temperature. This also verifies the results of Figure 10. The increase in the plasticity of the UVEF-treated sample is caused by an increase results of Figure 10. The increase in the plasticity of the UVEF-treated sample is caused by an increase in the plasticity of the UVEF-treated sample is caused by the ultrasonic vibration stress, which promotes the rejuvenation of the in the free volume induced by the ultrasonic vibration stress, which promotes the rejuvenation of the amorphous alloy. The ultrasonic thermal effect that causes a temperature increase does not increase the plasticity of the amorphous sample. This is because the maximum temperature rise caused by the plasticity of the amorphous sample. This is because the maximum temperature rise caused by the remaining thermal effect in this study is 270 C. which is far lower than the To of the Zr-based ultrasonic thermal effect in this

BMGs. Therefore, at low temperatures, an increase in the temperature cannot increase the free volume. The Mateultrials as onic 2020, vibration 13,x FOR PEERS tress REVIEW is the main reason for the increase in the free volume and the degree 120 fof 15 rejuvenation demonstrated herein [31].

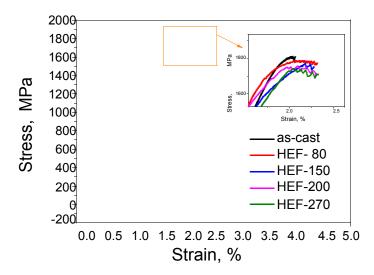


Figure Figure 11. 11 Fracture Stress -strain -strain curve curve of the as-cast-cast and and HEF HEF -treated samples samples samples.

3.6. Mathematical Model of the Relationship between Ultrasonic Amplitude And Free Volume

3.6. Mathematical Model of the Relationship between Ultrasonic Amplitude And Free Volume

The shear phase transformation zone (STZ) model [32] is widely used to describe the physical The shear phase transformation zone (STZ) model [32] is widely used to describe the physical process of plastic deformation of amorphous alloy materials. This STZ model is applicable whether the process of plastic deformation of amorphous alloy materials. This STZ model is applicable whether deformation is transient or time-dependent and uniform or nonuniform. Generally, shear deformation the deformation is transient or time-dependent and uniform or nonuniform. Generally, shear deformation is preferentially activated in areas with an increased free volume. Therefore, an increased free volume deformation is preferentially activated in areas with an increased free volume. Therefore, and leads to easier and more shear deformation, which in turn leads to better plasticity.

o easier and more shear deformation, which in turn leads to better plasticity.
increased free volume leads to easier and more shear deformation, which in turn leads to better It is deduced from this that the BMGs with an increased free volume show a decreased resistance

It is deduced from this that the BMGs with an increased free volume show a decreased resistance

free volume and increases the degree of rejuvenation of amorphous alloys.

to any form of deformation. The ultrasonic amplitude leads to a high-energy state, which increases To quantify the e ect of the ultrasonic amplitude on the rejuvenation degree of the amorphous the free volume and increases the degree of rejuvenation of amorphous alloys.

samples, the amount of free volume is used to characterize the rejuvenation degree of the amorphous alloys, so the amount of free volume in the UVEF-treated samples is theoretically calculated based on samples, the amount of free volume is used to characterize the rejuvenation degree of the amorphous alloys, so the amount of free volume in the UVEF-treated samples is theoretically calculated based on samples, the amount of free volume is used to characterize the rejuvenation degree of the amorphous

alloys, so the amount of free volume in the UVEF-treated samples is theoretically calculated based

The amount of free volume of the UVEF-treated sample can be obtained according to Equation (2), as shown in Figure 12. It can be found that the larger the amplitude of the UVEF-treated sample is, the $T T_0$ (4)

 DT_0

where v_{fe} is the equilibrium free volume; $k_t = 26.6 + 0.044T$ and l = 43.9 + 0.082T represent temperature- related parameters [33]; D is the brittleness index of the BMGs, which is taken as 18.5; and T_0 is the Vogel–Fulcher temperature, $T_0 = 2/3T_g$.

fitting is 0.99983, the mathematical model to obtain the ultrasonic amplitude (A) and free volume (v_f) is obtained as follows:

$$v_f = 7.9857 \ 0.20477A + 0.01216A^2$$
 (5)

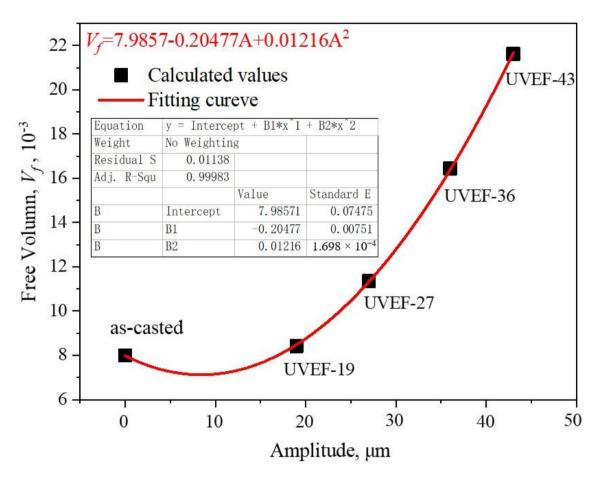


Figure 12. Theoretical model of the relationship between the free volume and amplitude of UVEF-treated samples.

As the amplitude increases, the amount of free volume increases sharply, indicating that the greater the ultrasonic energy is, the greater the degree of amorphous rejuvenation.

4. Conclusions

Compared with metallic glass rejuvenation methods such as elastic loading and ion radiation, UVEF has a short processing time (8s), will not cause damage, and is also controllable. It is found that the relaxation enthalpy increases and the range of the supercooled liquid region decreases when Zr based metallic glasses are UVEF-treated at room temperature. With the increase of amplitude, the free volume and the formability of the UVEF-treated sample increase, and the yield strength and elastic modulus decrease. Ultrasonic vibration stress is the main reason for the increase in the free volume of Zr-based BMGs.

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